

Quantitative estimation of secondary metabolites, in-vitro antioxidant, anti-inflammatory and anti-sickling activity of leaf of *Ficus virens* Aiton.

Research Article

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Abstract

Sickle cell anaemia is a severe global disease marked by oxidative stress and painful episodes. This study evaluates the antioxidant, anti-inflammatory, and anti-sickling properties of the ethanolic extract of *Ficus virens Aiton* leaves. Total phenolic (TPC), tannin (TTC), and flavonoid (TFC) contents were assessed spectrophotometrically, while antioxidant activity was measured using DPPH, anti-inflammatory activity via the HRBC method, and anti-sickling effects through Emmel and reversibility tests. The extract showed high TPC (79.84±2.77 μ g/mg GAE), TTC (10.51±0.62 μ g/mg GAE), and TFC (132.62±6.69 μ g/mg QE). It significantly reduced haemolysis (12.19±1.84% at 2000 μ g/ml) and exhibited strong antioxidant activity (89.34±0.56% DPPH scavenging at 75 μ g/ml, IC50: 22.76±1.01 μ g/ml). The extract also demonstrated a high sickle cell reversal rate (84.13±0.25% at 10 mg/ml after 150 min), comparable to phenylalanine (85.71%). These results highlight *Ficus virens Aiton* as a promising candidate for treating inflammation and sickle cell anaemia.

Keywords: Ficus virens Aiton, Anti-inflammatory, Anti-oxidant, Total phenolic content, Sickle cell anaemia.

Introduction

Sickle cell anaemia is a major concern throughout the globe. Thousands of children die due to its lethality every year around the world. As we know haemoglobin molecules undergo this change by replacing glutamic acid, a polar amino acid, with valine, a non-polar amino acid in position six (1,2,3). It is a result of the defective gene (mutation) on chromosome number 11. The mutant haemoglobin polymerises inside the red blood cells (RBCs) at low oxygen tension, resulting in a drastic reduction in the red cells deformability. Upon polymerisation and precipitation of sickle haemoglobin (HbS) within erythrocytes, sickle cells change from their normal spherical shape into one resembling a sickle. Single nucleotide substitution (Thymine in place of Adenine) allows HbS to polymerise when deoxygenated since valine can dock with complementary sites on adjacent globin chains (4). Herrick demonstrated sickle-shaped RBCs in human blood for the first time in 1910(5, 6).

It has been seen in sickle cell anaemia, the body suffers the problem of inflammation and oxidative stress. The pathophysiology of sickle cell disease (SCD) is characterised by inflammation (7). A tissue-resident cell of the innate immune system detects damaging

* Corresponding Author: Abhishek Kumar Pandey Assistant Professor, Department of Botany, Kalinga University, Naya Raipur, Chhattisgarh, India-492101 Email Id: abhishek.pandey@kalingauniversity.ac.in agents and activates nearby neutrophils. Activated monocytes migrate to the inflamed tissue, recruit inflammatory monocytes and promote proinflammatory conditions (8, 9). SCD is characterised by an increase in leukocyte count and activation of granulocytes, monocytes, and platelets (10, 7, 11). Different types of proinflammatory mediators derived from various sources including leukocytes, platelets, and endothelial cells such as tumour necrosis factor-alpha (TNF- α) and the interleukins (IL), IL-6, IL-1 β , and IL-8, are recorded in higher amount in sickle cell patients (12, 13,14). The multiplied production and release of proinflammatory cytokines can promote the vaso-occlusive procedure because of endothelial activation, erythrocytes, and leukocyte adhesion to vascular endothelial and endothelial cell apoptosis (15).

In Chhattisgarh, Sickle cell anaemia is a major concern and is one of India's most affected state (16). Indeed, the prevailing neighbouring states such as Maharashtra, Odisha, Jharkhand, Madhya Pradesh, and parts of Andhra Pradesh are also suffering the state of India. In Chhattisgarh, sickle-shaped haemoglobin is common in the central and southern regions of states which are Raigarh, Jashpur, Surguja, and Koriya. Chhattisgarh tribe with a high prevalence percentage of sickle cell haemoglobin is as follows: Halba in Rajnandgaon and Durg districts, Muria and Hillmaria in the Bastar district. Studies also show that scheduled caste and some OBC groups such as Sahu, Chandrakar, Kurmi, and Yadav in the south and central west of Chhattisgarh also have sickle-shaped haemoglobin. It's about the same ratio of a tribe in this area. Additionally, there are also 12 districts in Chhattisgarh, which are



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classified as the main area carrying SCD which includes Dantewada, Bastar, Kanker, Korba, Mahasamund, Rajnandgaon, Dhamtari, Kawardha, Bilaspur, Durg, Raipur, and Janjgir-Champa (16).

Indeed, out of the total population of the Chhattisgarh state that is 7.5 million (16) Gond tribe (20% of the total tribal population) is strongly affected by sickle cell disease. Sickle cell disease screened 359,823 subjects among 2,087 (99.7%) of the villages in Raipur District, Chhattisgarh State, India between October 2007 and June 2010. The sickle cell trait (sickle cell gene comes from a single parent) occurred in 33,467 (9.30%) and individuals with sickle cell disease (sickle cell gene comes from both parents) 747 (0.21%)(16). A previous study conducted in the Rajnandgaon district, also documented that 9.75% of the examined population carries a sickle cell gene (17). In another study, random sampling of 6088 people was done to test the sickle cell anaemia problem by slide test method, and a total of 249 (4.09%), people were found sickled positive. Further electrophoresis test was performed for all 249 of which 67 were found homozygous (HbSS) and 182 were found heterozygous (HbAS) positive in Rajnandgaon district (18).

Many plants were investigated for anti-sickling agents. Researchers around the globe are working on different medicinal plants and checking the potential of many phytoconstituents against this disease. The present study also aims to investigate the therapeutic potential of Ficus virens Aiton against inflammation, oxidative stress, and sickle cell anaemia. Ficus virens Aiton belongs to the Moraceae family found in the Indian sub-continent commonly known as white fig in English. It is a sacred tree in Indian methodology and is believed to be the dwelling place of lord Vishnu. In Ayurveda, this plant is mentioned in the name of Plaksha. According to Ayurveda, this plant has been used in urogenital diseases. Ethnobotanical bark decoction is used to strengthen the bone, bark powder is used to prevent bleeding and in the treatment of wounds. Fruits are used to treat diarrhoea (19). Fresh and young leaves of the plant are edible in some parts of China (20) Recent studies suggest that plants contain anti-viral (21), and anti-cancerous activity (22). The main objective of this study is to explore the antioxidant, anti-inflammatory, and anti-sickling activity of this plant (Ficus virens Aiton) and provide some alternative and effective plant-based drugs for the treatment of the above-mentioned problem.

Material and Methods Collection of plant material

The plant sample has been collected from the Raghuveer temple Jankikund Chitrakoot Satna, Madhya Pradesh where it grows enormously. The plant species is identified by its morphological features and confirmed by Dr. R.L.S. Sikarwar Taxonomic head of Deendayal Research Institute Chitrakoot Satna Madhya Pradesh, India. The plant material was washed with tap water two to three-time. The specimen voucher has been submitted to the Department of Botany, Kalinga University, and Naya Raipur for future reference (Specimen voucher No- APKU4).

Preparation of plant extract

Dried leaves were ground in the mixer and sieved with a $355\mu m$ IS sieve. This fine plant powder is used for Physico-chemical analysis. For extraction of plant extract, coarse powder has been prepared and loaded in an extraction chamber of the Soxhlet apparatus. Ethanol is used as a solvent

Physicochemical analysis

An array of physicochemical tests was conducted on plant powder, including the determination of alcohol-soluble extractives, water-soluble extractives, total ash, acid-insoluble ash, loss on drying, and pH determinations (23,24,25).

Determination of PH range

Using a standard glass electrode pH meter, the pH levels of various formulations containing watersoluble portions of *Ficus virens Aiton* leaf powder were measured at concentrations of 1% w/v (1g; 100 ml) and 10% w/v (10 g; 100 ml).

Determination of moisture content

To determine the loss on drying, approximately 1.0 g of *Ficus virens Aiton* leaf powder was placed on a precisely weighed moisture disc (measured using an electronic scale - Dawner). The sample was then dried in an oven at 105 °C for 3 hours, followed by cooling in a desiccator for 30 minutes before being weighed without delay. The weight loss was calculated as a percentage of the air-dried material.

Extractive Value

To determine the solubility of the extract, the sample was dissolved in different solvents, including water and ethanol.

Determination of total ash

An accurately weighed and previously ignited crucible was used to hold 2 grams of *Ficus virens Aiton* leaves powder. The dried powder was evenly spread in the crucible and ignited gradually by increasing the temperature to 450 °C for 5 hours in a muffle furnace until it turned white, indicating the absence of carbon. After cooling in a desiccator, the crucible was weighed, and the total ash content was determined in milligrams per gram of air-dried material.

Acid Insoluble Ash Value

After adding 25 mL of dilute HCl to the crucible containing total ash, the insoluble matter was collected on an ashless filter paper (Whatman number 41) and washed with hot water until the filtrate reached neutrality. The filter paper, with the insoluble matter, was then transferred to the original crucible, dried on a hot plate, and ignited until a constant weight was obtained. The residue was allowed to cool in a desiccator for 30 minutes before being weighed



promptly. The content of insoluble ash was calculated based on the air-dried plant material.

Phytochemical analysis: Qualitative

Plant extracts were subjected to qualitative chemical tests. The phytochemical screenings included tests for alkaloids, tannins, steroids, glycosides, flavonoids, saponins, carbohydrates, terpenoids, and proteins (23-25).

Test for Alkaloids:

To test for the presence of alkaloids in the leaves extract, approximately 15 mg of the extract was mixed with 6 ml of 1% HCL and stirred in a water bath for 5 to 6 minutes. The resulting solution was then divided equally into three parts.

1. One of the portions was subjected to Dragendorff's test by adding 1 ml of Dragendorff's reagent to 2 ml of the solution. The formation of an orange-coloured precipitate indicated the presence of alkaloids.

2. To perform Mayer's test, 1 ml of Mayer's reagent was added to a portion of the solution (2 ml), and the appearance of a cream-coloured precipitate indicated the presence of alkaloids.

3. The above solution was subjected to Wagner's test by adding a few drops of Wagner's reagent. The formation of a brown precipitate indicates the presence of alkaloids.

Test for Terpenoids (Salkowski test)

A reddish-brown colour indicates the presence of terpenoids when 100 mg of the extract is mixed with 2 ml of chloroform and 2 ml of concentrated H_2SO_4 added along the side of the test tube.

Test for Steroid

Around 1 mL of the crude extract was added to 10 mL of chloroform and 10 mL of sulphuric acid. The appearance of a bilayer, with a red upper layer and a greenish lower layer, confirms the presence of steroids.

Test for Tannins

A mixture was prepared by adding 0.5 g of leaf extract to 10 ml of distilled water, followed by the addition of a few drops of 5% ferric chloride. The presence of tannins is indicated by the formation of a black or blue-green precipitate.

Test for Saponins

The presence of saponins in 0.5 g of leaf extract was determined by shaking it with 10 ml of distilled water in a test tube and warming it in a water bath for 5 minutes. If foaming occurred, it indicated the presence of saponins.

Test for Glycoside

Anthraquinone glycoside (Born Trager's test)

The extract solution (1 ml) was mixed with 1 ml of 5% H₂SO₄ and boiled in a water bath, followed by filtration. Chloroform (equal volume to the filtrate) was added, and the mixture was allowed to stand for 5 minutes. Then, half of its volume of dilute ammonia

was added. The appearance of a rose-pink to red colour in the ammonium layer indicates the presence of anthraquinone glycosides.

Cardiac glycoside (Keller-Killiani test)

A mixture of 0.5 g of extract and 5 ml of distilled water was prepared, followed by the addition of 2 ml of acetic acid, some ferric chloride, and 1 ml of H_2SO_4 along the side of the test tube. Synthesis of cardiac glycoside is indicated by the appearance of a brown ring at the interface, followed by a violet ring.

Baljet test

Mix 2 mL of the extract with 1 mL of Baljet reagent (prepared by combining 95 mL of 1% picric acid and 5 mL of 10% NaOH). The appearance of an orange-yellow colour indicates the presence of cardiac glycosides (76).

Phytochemical analysis: Quantitative

Total Phenolic Content (TPC)

For the determination of TPC, the Folin-Ciocalteu method was applied using a spectrophotometer with some modification (Labtronics-2201) as first described by Singleton & Rossi in 1965 (26). First of all 1 mg/ml concentration solution of the plant extract was prepared. Now transfer 1 mL of this solution to a 25 ml volumetric flask. Then 9 ml of distilled water was added to this solution. Then 1ml Folin-Ciocalteu reagent was added to this solution and stirred well for five minutes and then 10 ml of 7% sodium carbonate. Finally, the final volume was made to 25 ml by adding distilled water. Similarly, different concentrations of standard solutions of Gallic acid 20, 40, 60, 80, and 100 µg/ml were prepared and absorbance was noted against the reagent blank at 750 nm on a UV-vis spectrophotometer after 90 minutes of incubation (27-30).

Total Tannin Content (TTC)

To estimate the total tannin content in plant extract, the Folin-Ciocalteu method was used (31-32). First of all, 0.1 ml of the plant extract was transferred to 10 ml of the volumetric flask. Then 7.5ml of distilled water was added to this solution. Now 0.5 ml of Folin-Ciocalteu reagent was added to this solution. After 5 minutes, 1 ml of 35 % Sodium carbonate was added and the mixture was shaken. A set of standard solutions of Gallic acid (20, 40, 60, 80, and 100 μ g/ml) was prepared in the same manner. The absorbance of the sample and standard solutions were taken at 725 nm with the help of a UV-visible spectrophotometer against the reagent blank after 30 minutes of the incubation period.

Total Flavonoid Content (TFC)

By using a spectrophotometer aluminium chloride assay, the total flavonoid content of leaf extract has been determined. 1 ml of plant extract has been taken in the 10 ml of the volumetric flask for this experiment. Further, we added 0.3 ml of 5% NaNO₂ after adding 4 ml of distilled water and shaking well.



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After five minutes, we added 0.3 ml of 10% AlCl₃ and 2 ml of 1M NaOH. Double distilled water was added to this mixture to make the final volume 10 ml. By following the same manner set of standards drug quercetin (20, 40, 60, 80, and 100 μ g/ml) has been prepared. Using a UV-visible spectrophotometer model LT-2201, absorbance measurements for the standard and test solutions were made at 510 nm after the 30-minute incubation period (33-34).

Anti-inflammatory activity: Human Red Blood Cell (HRBC) method

Preparation of Human Red Blood Cell (HRBC) Suspension

Fresh human blood has been ejected through a syringe from a volunteer. The detail and purpose of the studies have been told to the volunteer and written consent has been taken from the participant. 2 ml of human blood has been mixed with equal volume sterile Alsever solution (2 % dextrose, 0.8 % sodium citrate, 0.05% citric acid and 0.42 % sodium chloride is dissolved in 100ml of water). Centrifuge the mixture for 10 minutes at 3000 rpm, packed cells were washed three times with isosaline. The measured blood volume was used to create a 10% v/v isosaline solution.

Heat-induced Haemolysis

According to this method, membrane lysis was caused when the human blood cell membrane was induced with a hypotonic solution. If the extract/drug can stabilise the HRBC membrane, haemolysis can be prevented. First of all 1 ml of phosphate buffer (pH 7.4, 0.15M), has been taken in a test tube. Added 2 ml of hypo saline (0.36%), and 0.5 ml of HRBC suspension (10% v/v) then add 0.5 ml of plant extracts and various concentrations of the reference drug diclofenac sodium (50, 100, 250, 500, 1000, and 2000 μ g/ml) in different test tubes. The percentage of haemolysis was determined using a spectrophotometer with a 560 nm setting. (35-37).

The percentage of Haemolysis of RBC was calculated by using the following formula:

The Percentage of Hemolysis = $\frac{\text{Absorbance of test sample}}{\text{Absorbance of Control}} X 100$

The percentage of HRBC membrane stabilisation can be calculated as follows:

The Percentage of HRBC Membrane Stabilization = $100 - \left[\frac{Absorbance of test sample}{Absorbance of Control} X 100\right]$

Free Radical Scavenging Activity (DPPH assay)

The DPPH method was used to assess the ability of the ethanolic extract of *Ficus virens Aiton* leaves to scavenge free radicals. It has been checked if the plant sample scavenges coloured α , α -diphenyl- β picrylhydrazyl (DPPH) which colour fades after scavenging. Standard DPPH solution prepared by dissolving the crystal of DPPH) (1.3 mg/ml in methanol). 10 ml of methanol have been used to dissolve 10 mg of plant extract. One ml of this solution was placed in test tubes and diluted up to ten times using the same solvent. Stock solutions with concentrations of 10, 15, 25, 50, and 60 μ g/ml were obtained and placed in different test tubes in amounts of 0.10, 0.15, 0.25, 0.50, and 0.60 ml. Make the final volume up to 1 ml by adding the same solvent. 75 μ l of freshly prepared DPPH was added to all test tubes. The absorbance was measured at 517 nm using a Labtronics UV-visible spectrophotometer (model no. 2201). A set of ascorbic acids dissolved in water with the same concentration was used to create a control sample. The following equation was used to determine the percentage of radical scavenging activity in test and standard samples. Each experiment was carried out three times. The percentage of inhibition by different concentrations of test samples and standards was calculated by the following equation (38-40).

The Percentage of Inhibition = $\frac{A \text{ control} - A \text{ sample}}{A \text{ control}} X100$

A control = Absorbance of DPPH reagent alone A sample = Absorbance of DPPH reagent along with different concentrations of extracts.

Anti-sickling activity (Emmel Test)

Sickle cell blood has been collected from the 14year-old patient whose treatment is going in Sickle cell institute Raipur. Written consent has been taken from her parents. Sickle cell blood was diluted with 150mM phosphate-buffered saline (NaH₂PO₄ 30mM, Na₂HPO₄ 120mM, NaCl 150mM) and mixed with an equivalent volume of 2% sodium metabisulphite (Na₂S₂O₅). A drop from the mixture was spotted on a microscope slide in the presence or absence of anthocyanins extracts and covered with a cover slip. Paraffin was applied to seal the edges of the cover completely to exclude air (Hypoxia). (41-42)

The anti-sickling activity of plant extract has been evaluated through a reversal test. Firstly all the erythrocytes were reduced by using sodium metabisulphite and incubated for 2 hours. After two hours cells were examined under the microscope and the percentage of sickling was calculated by using the following formula.

The percentage of sickling= $\frac{Number \ of \ sickling}{Total \ number \ of \ cell} \ge 100$

If the sickling percentage is below 100 %, the time had been increased accordingly to obtain 100% sickling. In our experiment, a 100% sickling rate was obtained after 2 hours of incubation.

Reversibility Test

Different concentrations of plant extracts starting from 50 μ g/ml to 10 mg/ml had been prepared in saline solution (0.86% NaCl) and tested against sickle cells at different time intervals. One hundred fifty microliters of plant extract had been added to the 150 μ l of the previous solution (previously 2 hours incubated blood with sodium metabisulphite). The same procedure was done with the positive control (phenylalanine) and the negative control.



The change in the percentage of residual sickle cells over time was given by the following relationship

Percentage of residual sickle cell $=\frac{Sickle\ cell\ average\ at\ Tx}{Sickle\ cell\ average\ at\ T0}$

 $T_x=0, 30, 60, 90, 120$, and 150 minutes T_0 = temperature initial The percentage of reversibility of sickle cell was calculated by the given formula

> Total number of normal cells Total number of sickled cells X 100

Result and Discussion

Morphology and anatomy of the plant: Ficus virens Aiton (Hindi Name- Plaksha, English Name-White Fig)

Ficus virens Aiton is a large spreading tree like other plants of the Fabaceae family. The plant stem was shown in Figure 1 (a). Leaves are petiolate; alternate; simple; stipulate as it is shown in figure 1 (b). The inflorescence is catkin type. The flower is small inconspicuous, bracteates and bracteolate, actinomorphic, incomplete, unisexual, and hypogynous. The plant contains four perianths which are arranged in two whorls. Stamens are present in four to five numbers. Male flowers are opposite to the perianth. Tow carpel is present in female flowers which are simple, the ovary is superior type, and apocarpous and basal placentation is found. Leaf and stem sections were cut out by free-hand sectioning using a blade. The transverse section of the leaf and stem show the epidermis, cortex, and endodermis. The vascular bundle is endarch, collateral, and conjoint as it is shown in Figure 2. (a), (b) and SEM image of the plant stem shown in Figure 2 (c)

Leaves are petiolate and green in colour. At immaturity, they are reddish-brown in colour but when they mature they acquire green colour. Leaves are simple alternate spirals. Aerial roots are also present to support the main trunk of the plant body but they are not as hard as the banyan tree. Aerial roots are attached to the main stem and give strength to the main stem. The milky latex is also present which is very nutritious in nature.

They attract lots of insects and ants during bleeding and injury of the plant portion. Leaves of Ficus virens Aiton start from 8cm to 19cm long and 3 to 6 cm wide. The midrib is whitish. Stipules are less than 1cm long. It is a deciduous tree. Plant shades their leaves in February and new leaves emerge in March with colours of purple red and bronze, giving the tree a wonderful look. The colour transformation goes on till April. It acquires a height up to 30 meters. Flowers unisexual; inflorescence syconia, axillary, paired, 1-1.5 cm across, globose, often obconical; peduncle 1-6 mm long, slender, pubescent; basal bracts 3, ovate, acute, persistent; orifice plane, closed by 3 flat apical bracts in a disc 1 mm wide, internal bristles abundant, white, chaffy-vesicular; flowers of 4 kinds; male flowers ostiolar, sessile, in 2-3 rings; tepals 2-3, ovate, acute or



shortly gamophyllous, stamen 1, filament 0.5 mm; anther oblong, parallel; female flowers sessile; tepals 3-4, free, ovary superior, 1.2 mm, obovoid, sessile or stalked, red-brown, style filiform, tapering, gall flowers sessile or shortly pedicellate, tepals 3-4, reddish, spathulate to linear-lanceolate, free, ovary sessile or stalked, red-brown.

Physico-chemical analysis

The powdered drug was evaluated for its physicochemical analysis which includes a loss on drying total ash, acid insoluble ash, and water-soluble ash. The analysis of the physical and chemical properties of plant drugs is crucial for identifying impurities and mishandling during their preparation. A high amount of total ash can indicate improper handling or the presence of inorganic impurities, while acidinsoluble ash may suggest contamination with materials such as silica, earth, or sand. Comparing acid-insoluble ash with the total ash value of the same sample can help differentiate between natural ash variations and contaminants. Water-soluble ash represents the portion of the total ash that dissolves in water and is a useful indicator of the presence of water-soluble salts in the drug. Extractive values reflect the presence of polar or nonpolar compounds that can be extracted from plant material. Removing moisture from crude drugs is essential to prevent spoilage by bacteria and molds, as well as to avoid enzymatic destruction of active ingredients (43). The finding of all these parameters is given in Table 1.

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Table 1: Physico-chemical analysis of plant				
	Parameters	Outcome		
-				

Loss on drying (LOD) %	7.63±1.01
Solubility on water %	7.28±0.93
Solubility in Ethanol %	7.33±0.85
Total Ash %	11.46±1.23
Acid Ash Insolubility %	2.04±0.53
PH of 1% w/v formulation solution	6.5

Phytochemical analysis of plant extract

Preliminary phytochemical testing of the plants revealed the presence or absence of alkaloids. terpenoids, tannins, reducing sugar, saponins, and glycoside. Phytochemical screening is a useful tool that not only identifies the various constituents present in plant extracts but also determines which ones are present in higher quantities. Additionally, this screening process can aid in the discovery of bioactive compounds that have the potential to be used in the development of new and effective drugs (44). Previous research showed that alkaloids contain antiinflammatory (45), antimalarial (46), and antimicrobial activities (47). Similarly, the presence of steroids has demonstrated that plants should have insecticidal and antibacterial properties (48). Other secondary metabolites such as cardiac glycosides are used to treat congestive heart failure and cardiac arrhythmia (49). Table 2 shows the presence or absenteeism of phytochemicals in Ficus virens Aiton.

Table 2: Phytochemical screening of the Ficus virens Aiton

Test	Presence (+), Absence (-)				
Alkaloids					
(a) Dragendorff's test	+				
(b) Mayer's Test	+				
(c) Wagner's Test	+				
Terpenoids					
Salkowski Test	+				
Steroid Test	+				
Tannin	+				
Reducing sugar	+				
Saponins	+				
Glycoside					
(a) Anthraquinone Glycoside	-				
(b) Cardiac Glycoside (Baljet test)	+				
(c) Keller-Killiani test	-				

Total Phenolic Contents (TPC)

Total phenolic content in terms of equivalent to Gallic acid was quantified per μ g/mg of plant extract. By using the standard curve and equation derived from the graph i.e. y = 0.0026x - 0.0026, $R^2 = 0.9981$ (figure 3), the values of the total phenolic content of the test sample have been calculated. Total phenolic content (TPC) is recorded at 79.84±2.77 μ g/mg of GAE (Gallic acid equivalent). Previous research showed that the methanol fraction of *F. virens* contained a phenolic content of 1267.35 mg GAE/g dry extract (50). Another study conducted in Egypt found TPC only 2.50±0.715 mg/ gram of Gallic acid equivalent (51). Redox



Figure 4: Standard curve of Gallic acid for total tannin content



Figure 5: Standard curve of quercetin for total Flavonoid Content



properties attributed to phenolic compounds, which are significant constituents of plants, play a crucial role in their antioxidant activity (52).

Total Tannin Contents (TTC)

The total tannin content of the leaves of *Ficus* virens Aiton was calculated by using the formula y = 0.0094x + 0.0075, $R^2 = 0.9984$ (figure 4) extracted from the standard curve of Gallic acid for tannin estimation. The value of total tannic contents was expressed in terms of μ g of per mg of Gallic acid equivalent. Leaves of ethanolic extract of *Ficus virens Aiton* showed $10.51\pm0.63 \mu$ g/mg of Gallic acid equivalent. Numerous studies have demonstrated the antioxidant potential of major natural product constituents such as polyphenols, terpenes, and polyunsaturated fatty acids (53, 54)

Total Flavonoid Contents (TFC)

The total flavonoid content of the plant part was calculated by using the formula y = 0.0009x + 0.0013, $R^2 = 0.9984$ (figure 5). The values of TFC are expressed in terms of µg per mg of the equivalent of quercetin. The values of TFC were recorded at 132.63±6.70 µg/mg of quercetin equivalent (QE). Previous research showed that the methanolic extract of leaves of *F. virens* contained the maximum amount of flavonoids than other *Ficus* species such as *Ficus* benghalensis, *Ficus* religiosa, *Ficus* elastica. (1080.61 mg QE/g dry extract) (50). Secondary metabolites known as flavonoids exhibit antioxidant properties, and the effectiveness of their antioxidant activity is influenced by the quantity and arrangement of unbound hydroxyl groups (55).

Antioxidant activity of Ficus virens Aiton

DPPH is a free radical with an unpaired electron located on nitrogen, which exhibits a robust absorption peak at 517 nm. Its colour changes from purple to vellow when it undergoes electron-pairing with a radical scavenger, leading to the formation of reduced DPPH-H (56). The ethanolic extract of Ficus virens Aiton (EEFV) showed strong antioxidant potential in the DPPH method as shown in Figure 6. At 75 µg/ml, the highest antioxidant activity has been recorded it will scavenge 89.34±0.56% of DPPH at the same concentrations as ascorbic acid scavenges 59.13±0.98% so it will be quite better than ascorbic acid. IC_{50} of EEFV is observed at 22.76±1.01 µg/ml so it is almost half of the IC 50 of ascorbic acid which is 51.88±0.43 µg/ml. The percentage of inhibition of DPPH by different concentrations of the standard drug was given in Table 3 and the inhibition of DPPH by EEFV has been given in Table 4. A previous study showed that the

Figure 6: Antioxidant potential of *Ficus virens Aiton* with the comparison of Ascorbic acid through DPPH assay



antioxidant activity of *Ficus virens Aiton* leaf extract significantly is dependent on concentration. Young leaf extract of *F. virens var. sublanceolata* inhibits 97.68% ABTS radical at 0.5 mg/ml of extract concentration. *F. virens var. sublanceolata* and, *F. virens var. verins* required 0.34 mg/ml and 1.03mg/ml to scavenge 50% of the DPPH (57). Earlier reports have demonstrated scavenging activities of ethanol extract of *F. carica* L.

 Table 3: Percentage of inhibition of DPPH by

 different concentrations of Ascorbic acid

First reading	Second reading	Third reading	Mean				
26.1	24.03	27.32	25.82±1.66				
28.72	33.02	28.56	30.10±2.53				
30.05	31.15	32.04	31.08 ± 1.00				
36.1	32.35	35.12	34.52±1.94				
39.45	36.15	37.38	37.66±1.67				
52.15	47.05	50.12	49.77±2.57				
61.35	56.02	60.04	59.14±2.78				
	<i>First</i> <i>reading</i> 26.1 28.72 30.05 36.1 39.45 52.15 61.35	First reading Second reading 26.1 24.03 28.72 33.02 30.05 31.15 36.1 32.35 39.45 36.15 52.15 47.05 61.35 56.02	First readingSecond readingThird reading26.124.0327.3228.7233.0228.5630.0531.1532.0436.132.3535.1239.4536.1537.3852.1547.0550.1261.3556.0260.04				

Table 4: Percentage of inhibition of DPPH by
ethanolic extract of Ficus virens Aiton

Concentrat ions	First reading	Second reading	Third reading	Mean		
5µg/ml	36.8	34.56	34.23	35.20±1.40		
10 µg/ml	37.2	35.46	34.97	35.88±1.17		
15 μg/ml	46.53	43.64	43.85	44.67±1.61		
20 µg/ml	50.15	48.14	50.04	49.44±1.13		
25 μg/ml	52.35	50.85	49.68	50.96±1.34		
50 µg/ml	75.25	72.87	71.2	73.11±2.04		
75 µg/ml	90.56	88.21	89.25	89.34±1.18		

fruits and the methanol extract of *F. microcarpa* L. fil. leaves against ABTS+ and DPPH free radicals (58, 59).

The value of inhibition of DPPH by EEFV has been calculated by using the formula (y = 0.7942x + 31.392) and the value of standard drug inhibition was calculated by the given formula (y = 0.4734x + 24.772).

The present study shows leaf extract contains significant antioxidant activity in comparison to ascorbic acid. A lower IC_{50} value indicates higher antioxidant potential. The result also showed that plant contains high amounts of flavonoid and phenolic compounds which contributed to enhancing the antioxidant potential of plant parts. The redox properties of phenolic compounds are responsible for their antioxidant activity, which plays a crucial role in absorbing and neutralising free radicals, as well as quenching singlet and triplet oxygen. Additionally, phenolic compounds possess the ability to chelate metals (60).

Anti-inflammatory activity of Ficus virens Aiton

Ficus virens Aiton show significant antiinflammatory activity and an almost similar response to diclofenac sodium. At 50 µg/ml of the extract, 55.51 ± 2.63 % of haemolysis was observed which was further reduced up to 12.19 ± 1.84 % at 2000 µg/ml while in the case of diclofenac sodium; 45.25 ± 1.41 % of haemolysis was recorded at 50 µg/ml which was further reduced up to 9.57 ± 2.03 at 2000 µg/ml of concentrations. Figure 5 shows the reduction trend of haemolysis on adding the concentrations while Figure 8 shows the percentage of protection. Table 5 shows the haemolysis reduction data of ethanolic extract of *Ficus virens Aiton* (EEFV) and standard drug diclofenac sodium.

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Figure 7: Haemolysis inhibition of HRBC through diclofenac sodium and ethanolic extract of *Ficus virens Aiton* (EEFV)



Figure 8: Protection of lysis of HRBC in the presence of diclofenac sodium and ethanolic extract of the *Ficus virens Aiton*





Concentrations	Ethanolic extract of the <i>Ficus virens</i> <i>Aiton</i>	Standard drug (Diclofenac Sodium)
50 μg/ml	55.51±2.63	45.25±1.41
100 µg/ml	42.48±157	38.76±2.61
250 µg/ml	30.75±2.34	30.25±1.05
500 µg/ml	27.65±3.27	23.43±1.54
1000 µg/ml	21.5±2.19	17.12±2.36
2000 µg/ml	12.19±1.84	9.57±2.03

The anti-inflammatory activity of plant extract is evaluated through the HRBC suspension method. The ervthrocyte membrane is similar in structure to the lysosomal membrane, the stabilisation of the former suggests that the extract could potentially stabilise lysosomal membranes as well (61). Preventing the release of lysosomal constituents, such as bactericidal enzymes and proteases, from activated neutrophils is crucial for limiting the inflammatory response and reducing tissue inflammation and damage caused by their extracellular release. Thus, the stabilisation of the lysosomal membrane plays a vital role in achieving this goal (62). While the precise mechanism by which the extract stabilises the membrane is still unknown, hypotonicity-induced haemolysis is thought to result from the shrinking of cells due to the loss of intracellular electrolytes and fluid components through osmosis. The extract could potentially inhibit the

Figure 9: Percentage of the reversion rate in the presence of the ethanolic extract of *Ficus virens Aiton* at concentrations of 50 µg/ml







Figure 11: Percentage of the reversion rate in the presence of the ethanolic extract of *Ficus virens Aiton* at concentrations of 5 mg/ml









processes that stimulate or facilitate the efflux of these intracellular components (63, 64).

Anti-sickling activity of Ficus virens Aiton

Figures 9 to 12 show the anti-sickling activity of plant extract. The anti-sickling activity of *Ficus virens Aiton* has shown a significant reversal rate of sickle cell. The extract showed almost equal potential to phenylalanine. As the concentrations increased, the reversibility rate also increased in the same manner. At 50 µg/ml of concentrations of plant extract 43.62± 1.00 reversal rate has been recorded after 150 minutes of the

incubation period. The anti-sickling activity of plant extract increased with the enhancement of concentrations. The reversal rate has been increased when we treat sickle cell with 500 μ g/ml and goes up to 51.31±1.50%. When the concentrations increase ten times more it will show a 72.48±0.50 reversal rate and its further increase reached up to 84.13±0.25 % when sickle cells are treated with 10 mg/ml of concentrations and 150 minutes of the incubation period. All the values of reversibility of sickle cells at different concentrations and time periods are given in Table 6.

 Table 6: Percentage of reversal of sickle cells at different concentrations of *Ficus virens Aiton* and Phenylalanine at different time intervals

Time in Minutes	Fv* 50 μg/ ml	Pa* 50 μg/ml	Fv 500 μg/ ml	Pa 500 μg/ ml	Fv 5 mg/ml	Pa 5mg/ml	Fv 10 mg/ ml	Pa 10 mg/ ml
30	16.89 ± 0.65	18.67 ± 1.34	23.88±1.65	25.93±1.76	35.83 ± 1.78	31.17 ± 0.78	40.88 ± 2.00	45.95 ± 1.92
60	$23.7 \pm .085$	28.00±156	30.97±1.35	31.48±137	41.98±1.24	39.61±0.89	50.38±1.75	52.60±1.76
90	29.93±1.25	37.33±1.96	37.75±0.85	38.27±1.64	50.00±1.55	49.35±1.36	59.86±1.37	62.99±1.32
120	36.69±1.67	43.33±1.38	44.30±1.15	48.15±2.04	58.45±1.65	57.79±1.45	70.92±0.65	70.78±1.54
150	43.62±1.00	52.00±1.69	51.31±1.50	56.79±2.17	72.48±0.50	66.23±1.04	84.14±0.25	85.71±1.14
*Fv: Ficus virens Aiton, Pa: Phenylalanine								

The result of this study reported that plant extracts contain high amounts of phenolic and flavonoid content. In previous research, various flavonoids reported in plant extract including quercetin-3-O- α -D-arabinopyranoside, quercetin-3-O- β -D-galactopyranoside, kaempferol-3-O- α -Darabinopyranoside, and kaempferol-3-O- β -D-galactopyranoside (65). In addition to this, many phenolic compounds are also reported in the plant extract. It has been previously reported in the literature that flavonoids may act as potent anti-sickling agents. The flavonoids present in the extract may be responsible for the plant's anti-sickling potential (66, 67).

The phytochemical screening of extracts showed the presence of different secondary metabolites such as phenols, flavonoids, alkaloids, and saponins. Extracts might have anti-sickling properties due to an inhibitory effect on in vitro haemoglobin polymerization or structural modifications linked to haemoglobin's environment (68).

Biomolecules' ability to avoid in vitro polymerization is determined by one or more of the following factors: (a) Their tendency and efficiency to bind to the complementary contact region/site of deoxyHbS monomers (68, 69); (b) Modification of the HbS contact region by amino acid residues (70); (c) Stabilization of the R state of HbS molecules (71, 72, 70). In addition to inhibiting sickle cell haemoglobin polymerization, antioxidant molecules improved oxidant status in sickle erythrocytes in studies (73-75)

Conclusion

The study showed that plant extract contains significant antioxidant, anti-inflammatory, and antisickling potential. Many more in-vivo and clinical studies are required to confirm its effectiveness against these ailments. If the result will be promising in the clinical study, then the herbal drug can be formed using the plant extract. This will not only replace the NSAIDs and hydroxyurea but also attract more plant-based studies to treat different ailments and diseases. Future research also can be focused and more centric to find out the therapeutic molecules present in the extract. Moreover, it is necessary to conduct toxicity profiling and bioguided fractionation studies

Conflict of interest statement

The authors declare there was no conflict of interest.

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